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**Overview** 



1. Goal

- 2. What Pressure Swing Adsorption Is / Is Not
- 3. Why Use It?
- 4. Suggestions For Technical Evaluations





- Given this year's format, we're going to err on the side of keeping things simple.
- **<u>If</u>**, by the end of this presentation:
  - You know 10% more about PSA than you know right now,
     -And-
  - You will consider PSA for your next project, we will be pleased.



#### Pressure Swing Adsorption What PSA Is

- RENEWABLE NATURAL GAS
- Adsorption is a natural phenomenon where gas molecules stick to a solid surface (adsorbent), based on differences in their natural tendencies to stick to the adsorbent.
- Adsorption is used is to separate a mixture, usually into a product and byproduct.
- Pressure Swing Adsorption (PSA) uses <u>high pressure</u> to force adsorption, then <u>lower</u> pressure to force desorption (regeneration of the media).



There are several good sources of knowledge on adsorption; we have contributed directly to many of them.



What PSA Is



- To separate  $CO_2 + H_2O + H_2S + VOCs + Siloxanes + N_2 + O_2$  from  $CH_4$ , <u>2 Stages</u> are required.
- Stage 1 removes the contaminants that adsorb more strongly than methane (CO<sub>2</sub> + H<sub>2</sub>O + H<sub>2</sub>S + Siloxanes + VOCs).
  - They are depicted as <u>*red*</u> in above diagram = Byproduct or Waste.
- The  $CH_4 + N_2 + O_2$  pass through the adsorbent.
  - They are depicted as <u>blue</u> in above diagram = Product.



RNG Works 2020

What PSA Is



- Stage 2 treats the partly purified LFG ( $CH_4 + N_2 + O_2$ ).
- Stage 2 removes the contaminants that adsorb less strongly than methane (N<sub>2</sub> + O<sub>2</sub>)

   This is depicted as <u>blue</u> in above diagram = Product (RNG).
- The  $N_2 + O_2$  pass through the adsorbent
  - They are depicted as *grey* in above diagram = Byproduct or Waste.



What PSA Is Not

Adsorption is:



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- <u>Not Expensive</u> relative to other separation methods:
  - CO<sub>2</sub> separation from LFG via PSA typically <u>costs less</u> than competing technologies.
  - $\circ$  N<sub>2</sub> & O<sub>2</sub> separation via PSA is <u>expensive</u> relative to the CO<sub>2</sub> separation, but it's also a more difficult separation.
  - PSA is the only technology that can accomplish both the  $CO_2$  separation from  $CH_4$  and the  $N_2 + O_2$  separation from  $CH_4$ , in a practical manner.
- <u>Not Easy To Design</u> it requires experience, knowledge, and data. Achieving high-performance (product purity, methane recovery, efficiency, reliability, and economy) requires and expert!

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Why Use PSA?



Combination of Flow Rate & Contaminant Reduction

**Combination of Flow Rate & Contaminant Reduction** 

- While the raw numbers differ, the shapes of the curves are nearly identical for both stages.
- CAPEX:
  - Typically tied to flow rate & contaminant reduction.
  - Relative cost of <u>small</u> systems can seem high due to fixed costs (e.g., controls), but it scales gradually.
  - There are opportunities for significant Balance of Plant savings.
- OPEX:
  - Typically tied more to flow rate, so it's pretty linear.
  - Relatively low operating pressures (~100 psi) means relatively low overall HP. (Stage 1)

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**Technical Evaluation** 



When comparing process alternatives, we suggest a holistic approach:

- CAPEX
- OPEX
  - Horsepower (be sure to include feed and product compression requirements)
  - Media—life & replacement cost
- Recovery = Revenue
- Technical Support—types, availability, expected uptime
- Adaptability—how does the process handle changes in LFG flow rate, composition, and ambient conditions?





## **Thank You!**

If you have questions about this presentation, adsorption, or how ARI can help you, please contact:

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Stay Safe